NITRIFICATION POTENTIAL ASSAY AIR RESOURCES BOARD PROJECT

June 23, 1995

Preparation.

- 1. 0.2 M KH₂PO₄ Solution: Dissolve 2.722 g KH₂PO₄ in 100 mL H₂O.
- 2. 0.2 M K₂HPO₄ Solution: Dissolve 3.843 g K₂HPO₄ in 100 mL H₂O.
- 3. 50 mM (NH₄)SO₄ Solution: Dissolve 0.6607 g (NH₄)SO₄ in 100 mL H₂O.
- 4. Flocculant Solution: Dissolve 7.35 g CaCl₂ 2H₂O and 10.15 g MgCl₂ 6H₂O in 100 mL H₂O.
- 5. Nitrification Potential (NP) Solution: dilute 7 mL Solution 1, 18 mL Solution 2, 50 mL Solution 3, and 5.35 g NaClO₃ to 5 L, and pH to 7.2.
- 6. Label 4 Lachat tubes for each soil.

Method.

- 1. Measure 100 mL of NP solution into 250 mL Erlenmeyer flasks.
- 2. Weigh out 10 g fresh soil into each flask.
- 3. Place on orbital shaker, 60 cycles / min.
- 4. At 2, 8, 14, and 24 hours, transfer 10 mL of suspension from each flask to centrifuge tubes.
- 5. Add a few drops of flocculant solution to each sample.
- 6. Centrifuge at 3000 rpm for 10 min.
- 7. Pour off supernatants into Lachat tubes, cap, and freeze.
- 8. At the conclusion of 24 hours, sieve the residues from the soil slurries through a 2 mm mesh.
- 9. Place the rocks in weighing tins, and dry in a 70° C oven overnight.
- 10. Record the dry rock weights from the soil slurries, then discard the rocks.